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**CHEMISTRY**

**9701/32**

Paper 3 Advanced Practical Skills 2

**May/June 2017**

CONFIDENTIAL INSTRUCTIONS

**Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.**

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**The Supervisor's attention is drawn to the Supervisor's Report on page 7 which must be completed and returned with the scripts.**

If you have any queries regarding these Confidential Instructions, please contact Cambridge stating the Centre number, the nature of the query and the syllabus number quoted above.

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This document consists of **8** printed pages.

## Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the Question Paper should be attempted.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out. The following codes are used where relevant.

<b>C</b>	corrosive	<b>MH</b>	moderate hazard
<b>HH</b>	health hazard	<b>T</b>	acutely toxic
<b>F</b>	flammable	<b>O</b>	oxidising
<b>N</b>	hazardous to the aquatic environment		

The attention of Supervisors is drawn to any local regulations relating to safety and first aid.

Hazard Data Sheets should be available from your chemical suppliers.

### Before the examination

**1 Access to the Question Paper is NOT permitted in advance of the examination.**

**2 Preparation of materials**

Where quantities are specified for each candidate, they are sufficient for the experiments described in the Question Paper to be completed.

In preparing materials, **the bulk quantity for each substance should be increased by 25%** as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

**3 Labelling of materials**

Materials must be labelled as specified in these Confidential Instructions. Materials with an **FB** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FB** coded chemical is given in the Question Paper itself.

**4 Identity of materials**

It should be noted that descriptions of materials given in the Question Paper may not correspond with the specifications in these Confidential Instructions. **The candidates must assume the descriptions given in the Question Paper.**

**5 Size of group**

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

### Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), suitable eye protection and disposable gloves should be used where necessary.

3 *For each candidate*

1 × 50 cm<sup>3</sup> burette  
1 × burette stand and clamp  
1 × 25 cm<sup>3</sup> pipette  
1 × funnel (for filling burette)  
2 × 150 cm<sup>3</sup> or 250 cm<sup>3</sup> conical flask  
1 × white tile  
1 × 25 cm<sup>3</sup> measuring cylinder  
1 × 50 cm<sup>3</sup> measuring cylinder  
1 × 250 cm<sup>3</sup> beaker  
1 × foamed plastic (polystyrene) cup  
1 × thermometer (−10 °C to +110 °C at 1 °C)  
1 × glass rod  
8 × test-tube\*  
2 × boiling tube\*  
1 × test-tube rack  
1 × test-tube holder  
2 × teat/dropping pipette  
1 × Bunsen burner  
1 × heatproof mat  
1 × wash bottle of distilled water  
1 × pen suitable for labelling glassware  
1 × spatula  
paper towels

\*Candidates are expected to rinse and reuse test-tubes and boiling tubes where possible. Additional tubes should be available.

### Chemicals required

- 1 It is **especially important** that great care is taken that the confidential information given below does **not** reach the candidates either directly or indirectly.
- 2 It should be noted that descriptions of substances given in the Question Paper may not correspond with the specifications in these Confidential Instructions.
- 3 Particular requirements

hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
	<b>FB 1</b>	150 cm <sup>3</sup>	0.100 mol dm <sup>-3</sup> hydrochloric acid	Dilute 50.0 cm <sup>3</sup> of 2.00 mol dm <sup>-3</sup> HCl to 1 dm <sup>3</sup> . (See preparation instructions in the current syllabus for 2.00 mol dm <sup>-3</sup> HCl.)
	<b>FB 2</b>	150 cm <sup>3</sup>	0.110 mol dm <sup>-3</sup> sodium hydroxide	Dissolve 4.4 g NaOH [C] in each dm <sup>3</sup> of solution.
	<b>FB 3</b>	100 cm <sup>3</sup>	2.00 mol dm <sup>-3</sup> hydrochloric acid	See preparation instructions in the current syllabus.
[C]	<b>FB 4</b>	100 cm <sup>3</sup>	1.80 mol dm <sup>-3</sup> sodium hydroxide	Dissolve 72.0 g NaOH [C] in each dm <sup>3</sup> of solution.
	<b>FB 5</b>	5 cm <sup>3</sup>	1.00 mol dm <sup>-3</sup> ethanoic acid	Dilute 57.5 cm <sup>3</sup> of glacial (17.4 mol dm <sup>-3</sup> ) CH <sub>3</sub> COOH [C][F] to 1 dm <sup>3</sup> of solution.
	<b>FB 6</b>	5 cm <sup>3</sup>	2.00 mol dm <sup>-3</sup> hydrochloric acid	See preparation instructions in the current syllabus.
[C]	<b>FB 7</b>	10 cm <sup>3</sup>	2.00 mol dm <sup>-3</sup> nitric acid	See preparation instructions in the current syllabus.
	<b>FB 8</b>	10 cm <sup>3</sup>	0.10 mol dm <sup>-3</sup> copper(II) sulfate	Dissolve 25.0 g CuSO <sub>4</sub> .5H <sub>2</sub> O [MH] [N] in each dm <sup>3</sup> of solution.
[F][MH] [HH]	<b>thymol blue indicator</b>	5 cm <sup>3</sup>	thymol blue indicator	See preparation instructions in the current syllabus.
[C][MH]	<b>conc. HCl</b>	2 cm <sup>3</sup>	concentrated hydrochloric acid	Provide 2 cm <sup>3</sup> concentrated hydrochloric acid [C][MH] in a stoppered container.
	<b>FB 9</b>	5 cm <sup>3</sup>	0.10 mol dm <sup>-3</sup> ethylene diamine tetraacetic acid disodium salt dihydrate	Dissolve 37.2 g Na <sub>2</sub> (CH <sub>2</sub> N(CH <sub>2</sub> COOH)CH <sub>2</sub> COO) <sub>2</sub> .2H <sub>2</sub> O in each dm <sup>3</sup> of solution. This substance is commonly known as edta disodium salt.
[MH]	<b>sodium carbonate</b>	10 cm <sup>3</sup>	1.0 mol dm <sup>-3</sup> sodium carbonate	Dissolve 286.0 g Na <sub>2</sub> CO <sub>3</sub> .10H <sub>2</sub> O [MH] in each dm <sup>3</sup> of solution. OR Dissolve 106.0 g Na <sub>2</sub> CO <sub>3</sub> [MH] in each dm <sup>3</sup> of solution.
	<b>potassium iodide</b>	5 cm <sup>3</sup>	0.2 mol dm <sup>-3</sup> potassium iodide	Dissolve 33.2 g KI in each dm <sup>3</sup> of solution.
	<b>sodium thiosulfate</b>	5 cm <sup>3</sup>	0.2 mol dm <sup>-3</sup> sodium thiosulfate	Dissolve 49.6 g Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> .5H <sub>2</sub> O in each dm <sup>3</sup> of solution.

- 4 The reagents below should also be provided. Unless otherwise stated, each candidate should require no more than 10 cm<sup>3</sup> of any of these reagents. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the fact that such an arrangement may lead to contamination of reagents and enhance the opportunity for malpractice between candidates.

hazard	label	notes
	<b>dilute hydrochloric acid</b>	<b>See identity details and preparation instructions in the current syllabus.</b>
[C]	<b>dilute nitric acid</b>	
[MH]	<b>dilute sulfuric acid</b>	
[C] [MH] [N]	<b>aqueous ammonia</b>	
[C]	<b>aqueous sodium hydroxide</b>	
	<b>0.1 mol dm<sup>-3</sup> barium chloride or 0.1 mol dm<sup>-3</sup> barium nitrate</b>	
[N]	<b>0.05 mol dm<sup>-3</sup> silver nitrate</b>	
[MH]	<b>limewater</b>	
[MH]	<b>aqueous acidified potassium manganate(VII)</b>	
[MH]	<b>sodium carbonate</b>	
[F]	<b>magnesium ribbon</b>	Cut into 1 cm pieces (enough for two per candidate)

- 5 The following materials and apparatus should be available.

red and blue litmus papers, aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

### Responsibilities of the Supervisor during the examination

- 1 The Supervisor, or other competent chemist, **must, out of sight of the candidates, carry out the experiments in Question 1 and Question 2** and complete tables of readings on a spare copy of the Question Paper. This should be labelled 'Supervisor's Results' and show the Centre number and appropriate session/laboratory number.

This should be done for **each session** held and **each laboratory** used in that session, and **each batch** of solutions supplied.

**N.B. The Question Paper front cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.**

**It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.**

- 2 The Supervisor must complete the Supervisor's Report on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Supervisor's Report. A copy of the Supervisor's Report must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

### After the examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the barcode label provided.
- 2 A copy of the Supervisor's Results relevant to the candidates in 1.
- 3 A copy of the Supervisor's Report, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

### Colour blindness

With regard to colour blindness it is permissible to advise candidates who request assistance on colours of, for example, precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a Special Consideration Form.

## SUPERVISOR'S REPORT

**This form must be completed and returned in the envelope with the scripts, the Supervisor's Results, the Attendance Register and the Seating Plan.**

Centre number ..... Name of Centre .....

The candidate numbers of candidates attending each session were:

*first session*

*second session*

The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:

- any general difficulties encountered in preparation of materials;
- difficulties due to faulty apparatus or materials;
- accidents to apparatus or materials;
- assistance with respect to colour blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported directly to Cambridge on the Special Consideration Form.

**Report on any difficulties experienced by candidates.**

**Declaration (to be signed by the Supervisor)**

The preparation of this practical examination has been carried out so as to maintain fully the security of the examination.

Signed .....

Name (in block capitals) ..... (Supervisor)

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