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UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2010 question paper for the guidance of teachers

0654 CO-ORDINATED SCIENCES

0654/31

Paper 31 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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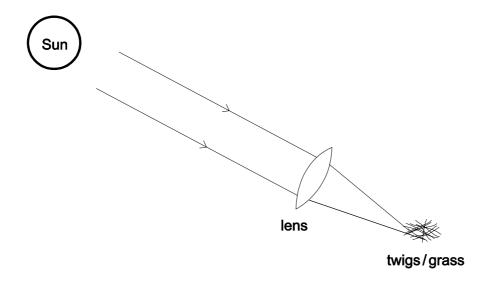
CIE is publishing the mark schemes for the May/June 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

	Page 2		ı	Mark Scheme: Teachers'		Syllabus	Paper
				IGCSE – May/June 20)10	0654	31
1	(a)	(i)	C ar	d D ;			[1]
		(ii)	A ar	d D ;			[1]
		(iii)	whe	ns and closes ; n atrium contracts valve is pushed on n ventricle contracts valve is pushe			[max 2]
	(b)	idea	a tha	gen (in right side of heart in fetus) it is a mix of oxygenated blood dy tissues);		and deoxygenated	[2]
	(c)	(i)	haer	noglobin ;			[1]
		(ii)	prote	ein ;			[1]
		(iii)	iron	;			[1]
		(iv)		I particles/not made of large moled an be absorbed as they are ;	cules ;		[2]
		(v)		espiration/to combine with glucose lease energy/to provide energy;	•		[2]
							[Total: 13]
2	(a)	(i)	caus in sk	es, skin cancer/eye damage/burr in ;	ns/mutation in sk	kin / damage to DNA	[1]
		(ii)	prote	ective clothing/sun block;			[1]
	(b)		eed = 00 m/) distance/time ; s ;			[2]
	(c)	•		um =) mass × velocity ; 0 × 60 = 24 000 000 kg m/s ;			[2]
	(d)	(i)		ymbols correct ; ymbols connected in series ;			[2]
		(ii)	6V;				[1]
	(e)	fabr	ric ga	transfer ; ins electrons / tent loses electrons / n between surfaces ;	or vice versa ;		[3]

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(f) two straight parallel rays drawn entering the lens; two straight rays brought to a focus at the twigs/grass; arrows correctly shown;





[Total: 15]

3 (a) ionising;

removes electrons;

damages DNA/mutation;

effect (e.g. cancer/burns/radiation sickness);

[max 3]

(b) (i) <u>nuclei</u> split/<u>nuclear</u> fission;

[1]

(ii) nuclear/radioactive/toxic waste; problems of disposal/storage;

p. c.

security of fissionable/radioactive material;

use in terrorism;

or

accident/malfunction;

effect of radioactive materials on environment/humans;

[max 2]

[Total: 6]

4 (a) (i) reaction is exothermic/heat was given off;

[1]

(ii) temperature falls (after 25 cm³ of acid added);

so no further (exothermic) reaction/all alkali used up;

[2]

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(b)	b) (i) moles of A $((25.0/1000) \times 0.2 =) 0.005$;				
	 moles KOH ((20.0/1000) × 0.5 =) 0.01; (allow 1 mark if the same error in converting to dm³ is made in each calculation, e.g. if left in cm³ answers are 5 and 10) (ii) (0.5) (no mark) [e.c.f. from (i) provided answer is half the KOH moles] because the number of moles of acid must be half the number of moles of KOH / owtte / or relevant working; 				[2] :h
	(iii)	H ⁺ +	$OH^- \rightarrow H_2O$ (all correct for 2 marks, two of the three	e for 1 mark) ;;	[2]
(c)	(i)	elec	trolysis ;		[1]
	(ii) plate, has a negative charge/is negative, and potassium ions, are positively charged/are positive; opposite charges attract/potassium ions move towards the plate;				
		•	ssium ions gain electrons from the plate; ssium ions, discharged/gain one electron/become	atoms ;	[max 3]
					[Total: 12]
					-
5 (a)	foai	m, sto	r, is a poor <u>conductor</u> ; ops <u>convection</u> of air/traps air; reflected by, shiny surfaces/foil/metal;		[3]
(b)	(i)		o mark) s ratio 2:1 ;		[1]
	(ii)		er can conduct electricity ; ger of electrocution ;		[2]
(c)	(i)	prod	ent (flows in circuit) ; luces (electro)magnet ; gnet) attracts iron bolt ;		[3]
	(ii)	•	no mark) ninium is not magnetic/not attracted to electromagne	et;	[1]
	(iii)		– no mark) an electromagnet (so still attracts bolt) ;		[1]
	(iv)	more	e coils/bigger voltage/bigger core ;		[1]
					[Total: 12]

6	(a)	(i)	ammonium/NH ₄ ⁺ ;	[1]
		(ii)	shortage of something in the soil; nitrogen/nitrate, needed for making, protein/amino acids; proteins for growth;	
			detail, e.g. more cells/more cytoplasm; correct ref. to function of P or K;	[max 3]
		(iii)	wheat – little/no, difference; potatoes – greater, with manure + bacteria/in plot B ; 10.50 tonnes (per hectare per year) (greater);	[3]
		(iv)	manure contains plant and animal waste e.g. proteins/urea; which needs to be, broken down/decomposed (by bacteria); to produce, ammonia/nitrates/something that can be used by plants; reference to nitrification/nitrifying bacteria;	[max 2]
			Total and the mention of the many many many many many many many many	[max 2]
	(b)	plaı	nulates growth of, algae/plants ; nts/algae, die ;	
		whi	on by bacteria / decomposers ; ch respire (aerobically) ;	
		bac	teria use oxygen ;	[max 3]
				[Total: 12]
7	(a)	(i)	glucose;	[1]
		(ii)	protein; only proteins contain, S/sulfur;	
			only proteins contain, N/nitrogen;	[3]
	(b)	(i)	molecules have only weak forces between them; molecules/particles, can move past one another easily; therefore (solid) nylon, melts / becomes a liquid, when heated / it enters	
			the hot container; molten nylon can be pumped (through small holes);	
			molten threads solidify when cooled; strong forces between molecules when solid;	[max 3]
		(ii)	doesn't melt (on contact with hot containers);	
			molecules cannot move past one another; because strong bonds hold polymer chains/crosslinks; [clear diagram could score crosslink mark]	[max 2]
				[Total: 9]

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8 (a) A to retina;

B to optic nerve;

C to iris;

(b) ciliary muscles, contract/get shorter;

loosen (tension on) (suspensory) ligaments;

lens more rounded/fatter;

more refraction/shorter focal length;

light (rays) brought to a focus on the retina;

[max 3]

[3]

(c) cystic fibrosis/sickle cell anaemia/thalassaemia/other; statement as to whether allele is dominant or recessive; (above examples are all recessive. Huntington's is dominant)

if recessive

both parents must have allele for offspring to inherit disease/are heterozygous; parental genotypes and offspring genotypes shown/1 in 4 chance of offspring having disease;

or

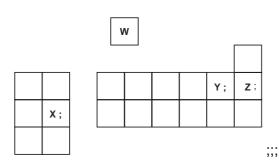
if dominant

only one parent needs to have allele for offspring to inherit disease; parental genotypes and offspring genotypes shown / 1 in 2 chance of offspring having disease;

[max 3]

[Total: 9]

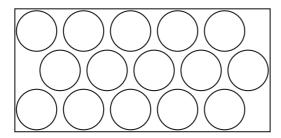
9 (a)



[3]

(b) (i) atoms all same size arranged in regular lattice; e.g.

[1]



(ii) reference to delocalised electrons;movement of charge/electrons;

[2]

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(c) (i) oxidation/reaction with oxygen (from air)/formation of metal oxide; reference to the, hot/molten, metal; [2]

(ii) three shells with 18 electrons; arranged 2,8,8; [2]

(iii) outer shell is complete; does not need to, lose/gain electrons, (by reaction)/owtte; [2]

[Total: 12]