www.xrenepabers.com

## **UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2010 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/31

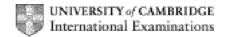
Paper 3 (Extended Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the October/November 2010 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



	Page 2			yllabus	Paper
			IGCSE – October/November 2010	0654	31
1 (	(a)	(i)	hydrogen;		[1]
		(ii)	$H^{+}$ ; allow $H_{3}O^{+}$		[1]
(	(b)	(i)	acid concentration; temperature; degree of agitation; <i>allow</i> size of test-tube		[max 2]
		(ii)	time taken for gas to fill test-tube was greatest;		[1]
		(iii)	rate is lower (with single piece); surface area (of single piece) is lower; fewer collisions per second/lower collision frequency (bet (atoms in) metal (surface));	tween acid and	d [3]
(	(c)	(i)	Mg + 2HC $l \rightarrow \text{MgC}l_2 + \text{H}_2$ ;; (formulae then look for balanced	i)	[2]
		(ii)	reference to the (granular) resin (beads); magnesium ions, removed / stick to the resin; sodium / hydrogen, ions (detach from resin and) enter the wat	er;	[max 2]
					[Total: 12]
2 (	(a)		ctrical energy to chemical energy ; so some of) electrical energy to heat ; <i>ignore light</i>		[2]
(	(b)		s distortion ; s interference ;		[2]
(	(c)	(i)	zero ; ignore units		[1]
		(ii)	force = 1.2/0.03; <i>allow ecf from</i> (i) = 40 N;		[2]
			rate of change of momentum slower larger force from concrete / smaller force from carpet; stopping force acted / energy transferred, over a longer period	I of time ;	[2] [Total: 9]

3	(a) (i)	<ul> <li>X sensory (neurone);</li> <li>Y relay / intermediate (neurone); A association / connector (neurone)</li> <li>Z motor / effector (neurone); [3]</li> </ul>
	(ii)	brain / spinal cord ; allow suitable named parts, e.g. medulla, grey matter [1]
	` ,	muscle; up / any other suitable response; [2]
	(c) (i)	contains amylase; changes starch to maltose / sugar; [2]
	(ii)	to produce small molecules (from large ones); so that the (small) molecules can be absorbed; pass through gut wall / move into the blood; so they can be used by cells; [2 max]
	(iii)	curve rises then falls ; peak between 30 °C and 40 °C ;  [2]
		[Total: 12]
4	(a) (i)	C <sub>8</sub> H <sub>18</sub> ; [1]
	(ii)	(octane) + oxygen - carbon dioxide + water  LHS; RHS; must be words – but allow one mark for completely correct balanced equation [2]
	(b) (i)	5;
	(ii)	three shared pairs; one non-bonding pair on both atoms; [2]
	(iii)	very strong bond (between the atoms); much energy needed to break bond / insufficient energy to break the bond; [2]
	(c) (i)	high strength, for safety/resist breakage/because high forces on airframe in flight; low density, to reduce weight/reduce fuel cost; [max 2]
	(ii)	$A_r$ of aluminium = 27; mass of aluminium = 1.73 × 27 = 46.74(g); allow other methods of working percentage in duralumin = $(46.74 \div 50.00) \times 100 = 93.4(2)\%$ [3]
		[Total: 13]

Mark Scheme: Teachers' version IGCSE – October/November 2010

Syllabus 0654 Paper 31

Page 3

5	(a)	(i)	0.47 A ;	[1]
		(ii)	resistance = voltage / current ; = $6/0.47 = 12.8 \Omega$ ; e.c.f.	[2]
	(b)	(i)	magnets repel ;	[1]
		(ii)	iron bar attracted to magnet ;	[1]
	(c)	(i)	magnetic field produced by current flowing through aluminium foil; this interacts with, other magnetic field / stronger magnet;	[2]
		(ii)	increase current / voltage ; increase magnetic field / stronger magnet ; ignore bigger magnet	[2]
	(d)		orking, e.g. $N_s = N_p \times V_s / V_p = 100 \times 35000 / 240$ 14 583 (allow correct rounding to 2 significant figures)	[2]
				[Total: 11]
6	(a)	(i)	grasses / other plants $\rightarrow$ white-tailed deer $\rightarrow$ wolves ;	[1]
		(ii)	wolves secondary consumers	
			white-tailed deer primary consumers	
		gras	producers	
			three <u>rectangles</u> drawn as above; each labelled with name of organism; each labelled with name of trophic level;	[3]
		(iii)	energy lost along food chains; (approx.) 90% lost/only 10% passed on; less energy available for, higher trophic levels/for wolves;	[2]
	(b)	(i)	ref. to <u>limiting factors</u> ; e.g. not enough food/more disease/competition for space;	[2]
		(ii)	maintain biodiversity; idea that loss of one species affects others in ecosystem; ethical/moral reason/ref. tourism/scientific research;	[2 max]
				[Total: 10]

Mark Scheme: Teachers' version IGCSE – October/November 2010

Page 4

Paper 31

Syllabus 0654

7	(a)	(i)	coloured compounds or variable, valency / ion charge / oxidation state;	[1]
		(ii)	Cu <sup>+</sup> ; working shows (or heavy implication of) need for charge balance; [reject unexplained "criss-cross" diagrams]	[2]
	(b)	(i)	anode labelled ; electrolyte labelled ;	[2]
		(ii)	copper chloride ; must be name, not formula	[1]
		(iii)	hydrogen ; must be name, not formula	[1]
		(iv)	cathode gas is hydrogen; so reactive metal present could be potassium; bromide ions negative so go to anode; bromine is orange (and would form from bromide and anode);	[max 2] [Total: 9]
8	(a)	•	vides / is, energy ; allow carbon dioxide to combine with water ; A to split water molecules	[2]
	(b)	(i) (ii)	place leaf in boiling water; place in hot alcohol (alcohol should be heated in a water bath for safety); dip in water (to soften); add iodine (solution);  area covered by paper shown on diagram; orange-brown where paper was, blue-black elsewhere;	[4]
	(c)	dur	pire all the time; ing daylight, photosynthesise <u>more</u> than they respire; piration takes in oxygen and produces carbon dioxide, photosynthesis vice	[2] [3]

Mark Scheme: Teachers' version IGCSE – October/November 2010

Page 5

Syllabus 0654 Paper 31

Page 6	Mark Scheme: Teachers' version	Syllabus	Paper
	IGCSE – October/November 2010	0654	31

9 (a) (force =) mass × acceleration; A weight = mass × gravity = 9.8 × 2 = 19.6 N; [2]

(b) (i) KE =  $\frac{1}{2}$  mv<sup>2</sup>; speed is 40 m/s; =  $\frac{1}{2}$  × 2 × 1600 = 1600 J; [3]

(ii) distance = under graph / other suitable working; height = 80 m; [2]

(c) (i) density = mass/volume; =  $2000/700 = 2.86 \text{ g/cm}^3$ ; [2]

(ii) use, displacement can / measuring cylinder / graduated container; place object in and measure, displaced water / difference in volume; [2]

(d) (i) Geiger counter / Geiger-Müller tube / any other suitable ; [1]

(ii) ionisation within cells/damages cells/kills cells/damages DNA/causes mutation/radiation burns/cancer/radiation sickness; [1]

[Total: 13]