

**CAMBRIDGE INTERNATIONAL EXAMINATIONS**

Cambridge Ordinary Level

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## **MARK SCHEME for the May/June 2015 series**

### **5090 BIOLOGY**

**5090/61**

Paper 6 (Alternative to Practical), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

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Mark schemes will use these abbreviations:

- ; separates marking points
- / alternatives
- () contents of brackets are not required but should be implied
- R reject
- A accept (for answers correctly cued by the question, or guidance for examiners)
- Ig ignore (for incorrect but irrelevant responses)
- AW alternative wording (where responses vary more than usual)
- AVP alternative valid point (where a greater than usual variety of responses is expected)
- ORA or reverse argument
- underline actual word underlined must be used by candidate (grammatical variants excepted)
- max indicates the maximum number of marks that can be given
- + statements on both sides of the + are needed for that mark

Question	Expected answers	Additional guidance	Marks												
1 (a) (i)	Benedict's (or Fehling's) solution ; heat ; <u>blue</u> + green / yellow / orange / red ; reference to use of a water bath / eye protection ;	R if add HCl / neutralise with NaOH  A boil / warm  accept in (i) or (ii)	[4]												
(ii)	biuret reagent ;  blue + purple / lilac ;	A sodium / potassium hydroxide + copper(II) sulphate	[2]												
(b) (i)	<table border="1" style="width: 100%; border-collapse: collapse;"> <thead> <tr> <th style="width: 50%;">concentration / g per dm<sup>3</sup></th> <th style="width: 50%;">time taken / s</th> </tr> </thead> <tbody> <tr> <td style="text-align: center;">0.0</td> <td style="text-align: center;">230</td> </tr> <tr> <td style="text-align: center;">0.2</td> <td style="text-align: center;">200</td> </tr> <tr> <td style="text-align: center;">0.4</td> <td style="text-align: center;">150</td> </tr> <tr> <td style="text-align: center;">0.6</td> <td style="text-align: center;">50</td> </tr> <tr> <td style="text-align: center;">0.8</td> <td style="text-align: center;">30</td> </tr> </tbody> </table>	concentration / g per dm <sup>3</sup>	time taken / s	0.0	230	0.2	200	0.4	150	0.6	50	0.8	30	one mark per column if all numbers correct  if units included in table max. 1  if conc. not recorded in ascending or descending order then max. 1	[2]
concentration / g per dm <sup>3</sup>	time taken / s														
0.0	230														
0.2	200														
0.4	150														
0.6	50														
0.8	30														
(ii)	<ol style="list-style-type: none"> <li>1. concentration on x-axis, time on y-axis, both axes fully labelled ;</li> <li>2. suitable scales: linear + minimum size specified ;</li> <li>3. all points plotted correctly ;</li> <li>4. points neatly joined by ruled lines ;</li> </ol>	A conc g / dm <sup>3</sup> , t / s  ± ½ small square (1 mm on grid)  R if line extrapolated	[4]												
(iii)	40 (seconds) ;	accept figure consistent with graph	[1]												

Question	Expected answers	Additional guidance	
(iv)	reference to inverse relationship / as conc. increases time taken decreases ;	<b>Ig</b> inversely proportional	[1]
(v)	reference to use of thermometer (to check temperature) ;  reference to method of controlling temperature, e.g. adding hot water, use of a water bath, means of heating water ;	<b>A</b> use of thermostat to measure temperature / device that is set at particular temperature  <b>Ig</b> lagging	[2]
(c)	min. 3 pH values used ;  reference to use of acid / alkali ;  volume of milk / enzyme / concentration of enzyme kept the same ;  temperature kept constant ;  coagulation time at each pH recorded ;  repeat + mean ;	<b>A</b> buffers  <b>A</b> stated volumes, e.g. 10 cm <sup>3</sup> milk  <b>A</b> stated temperature < 50 °C	[max 4]
<b>[Total 20]</b>			
2 (a)	P and Q only drawn with clear, clean continuous lines and no shading ;  cells P and Q drawn to correct scale (approx. 6 cm diameter) ;  cell walls indicated by double line ;  indication of chloroplasts in cell ;	tolerance 5 – 7 cm  either P or Q must show complete cell wall  not more than 8 chloroplasts in one cell	[4]
(b) (i)	increases until 1400 ;  to (a maximum of) 25 (g per hour);  then decreases ;		[3]
(ii)	(increases by) 23 g per hour ;		[1]

Question	Expected answers	Additional guidance	
(c)	cobalt chloride / <u>anhydrous</u> copper sulphate ;  colour change correctly described ;	blue to pink for cobalt chloride white to blue for copper sulphate	[2]
<b>[Total 10]</b>			
<b>3 (a) (i)</b>	35 ;  mm ;	<b>A</b> 34 – 36	[2]
(ii)	(35) ÷ 0.5 / actual = image ÷ magnification ;  correct answer with <u>units</u> ;	<b>A</b> multiplication by 2 <b>A</b> ecf from incorrect measurement in 3(a)(i)  34 = 68 mm / 35 = 70 mm / 36 = 72 mm	[2]
<b>(b) (i)</b>	widths totalled / <b>AW</b> ;  divided by the number in sample / 10 ;		[2]
(ii)	increase sample size ;  repeat with different species ;	<b>lg</b> repeat and calculate mean	[max 1]
(iii)	less light available in shade ;  (plants need) increased surface area ;  to trap more light ;  (for) <u>photosynthesis</u> ;		[max 3]
<b>[Total 10]</b>			